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Chemistry Sample Papers. Question.1. What is chromatography ? Answer. It is technique for rapid and efficient separation of components of a mixture and purification of compounds. It is based on differential migration of the various components of a mixture through a stationary phase under the influence of a

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moving phase.

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The paper strip after running of the solvent and drying is called chromatogram. What are the moving and stationary phases in paper chromatography? Both phases are liquid. The solvent is the mobile on moving phase while the water absorbed in the pores of the paper acts as a stationary phase.

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Viva Voce. Resources. Feedback. 1) Name the scientist who introduced chromatographic technique? Albert Einstein. Michael Faraday. Mikhail Tsvet. Isaac Newton. 2) What is the basic principle of paper chromatography? Difference in the boiling points of the components. Differential partitioning of components. Difference in the solubility of the ...

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Generally solvents having low viscosities are employed in chromatography. This is due to the fact that the rate of flow of a solvent varies inversely as its viscosity. 4. Name some chromatographic techniques. Paper chromatography, column chromatography, thin layer chromatography, gas chromatography. 5.

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a) Ascending paper chromatography b) Descending paper chromatography c) Radial paper chromatography d) Ascending – descending chromatography. Answer: c.

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Explanation: In Radial paper chromatography, the mobile phase moves horizontally over a circular sheet of paper. Separation takes place based on partition. 12. Liquid chromatography can be performed in which of the following ways? a) Only in columns b) Only on plane surfaces c) Either in columns or on plane surfaces d) Neither in columns ...

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level of the liquid in each beaker? What caused the dots of ink on the papers B and C to spread upwards?

### **Separating Mixtures – Exam Questions**

Paper chromatography involves the placing of a substance on a piece of filter paper and placing the edge of the paper in a liquid which, when travelling up the paper, will separate the components of the substance. For analysing some substances, like amino acids, two solvents are used.

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